

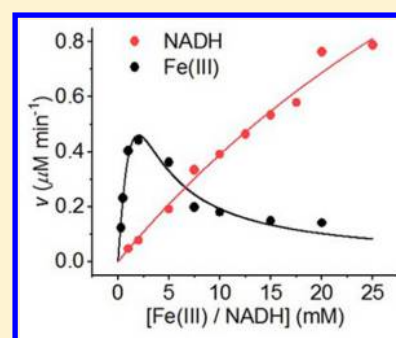
Steady-State Kinetics of α -Synuclein Ferrireductase Activity Identifies the Catalytically Competent Species

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S Supporting Information

ABSTRACT: α -Synuclein (α -syn) is a cytosolic protein known for its association with neurodegenerative diseases, including Parkinson's disease and other synucleinopathies. The potential cellular function of α -synuclein may be of consequence for understanding the pathogenesis of such diseases. Previous work has suggested that α -synuclein can catalyze the reduction of iron as a ferrireductase. We performed a detailed analysis of the steady-state kinetics of recombinant α -syn ferrireductase activity and for disease-associated variants. Our study illustrates that the ferrireductase activity we observed is clearly commensurate with bona fide enzyme activity and suggests a mechanistic rationale for the activity and the relationship to cellular regulation of the pool of Fe(III) and Fe(II). Using cell-based studies, we examined the functionally active conformation and found that the major catalytically active form is a putative membrane-associated tetramer. Using an artificial membrane environment with recombinant protein, we demonstrate that secondary structure folding of α -synuclein is insufficient to allow enzyme activity and the absolute specificity of the tertiary/quaternary structure is the primary requirement. Finally, we explored the steady-state kinetics of a range of disease α -synuclein variants and found that variants involved in neurodegenerative disease exhibited major changes in their enzymatic activity. We discuss these data in the context of a potential disease-associated mechanism for aberrant α -synuclein ferrireductase activity.



α -Synuclein (α -syn) is a member of a family of proteins termed the synucleins.^{1,2} It has been associated with a range of neurodegenerative diseases unified under the name of synucleinopathies, chief of which is Parkinson's disease (PD).^{3,4} Other synuclein family members include β -synuclein (β -syn) and γ -synuclein. α -syn and β -syn are colocalized in neurons,⁵ but α -syn has gained more attention, because of its potential to aggregate and form Lewy bodies that are seen in the synucleinopathies.⁶ α -syn is a 14 kDa protein that has been most commonly described as a monomeric, unstructured protein of the cytosol. However, there is considerable evidence that it can associate with the plasma membrane and gain helical content.^{7–10} There is also evidence suggesting that some amount of the cellular form can form tetramers that are rich in helical content.¹¹ The function of α -syn in cells remains unclear. Knockout of the protein in transgenic mouse lines does not alter viability or brain architecture but has mild effects on dopamine release.¹² There are some reports that α -syn knockout alters behavior.¹³

Various molecular studies have suggested a range of functions for α -syn. As the protein is expressed in dopaminergic neurons, it has been suggested that the protein regulates dopamine release.¹⁴ In particular, α -syn is thought to be involved in SNARE-dependent vesicle fusion leading to neurotransmitter release.¹⁵ This is supported by studies that suggest α -syn is localized to the presynaptic terminals,¹⁶ but α -syn has also been found in the nucleus. In this case, it has been

suggested to have a role in gene transcription via inhibiting histone acetylation.¹⁷ Other studies have suggested that α -syn is localized to the mitochondria, where it may be involved in mitochondrial function.¹⁸ α -syn has been suggested to bind metals, including copper and iron. In this regard, α -syn has been suggested to catalyze the reduction of iron.¹⁹

Ferrireductase (FR) activity in cells is vital to ensure the availability of Fe(II) for a range of cellular processes and particularly those enzymatic activities that require Fe(II) as a cofactor. In dopaminergic neurons, such as those in the substantia nigra, lost during that pathogenic advance of PD, Fe(II) is required as a cofactor for tyrosine hydroxylase involved in the synthesis of dopamine.²⁰ PD is also associated with disturbances in brain iron homeostasis.^{21,22} Recently, it has been suggested that α -syn can act as a ferrireductase.¹⁹ Therefore, α -syn, which is highly expressed in the substantia nigra, could serve a role in the regulation of the balance of Fe(II) to Fe(III) in this key region.

The previous evidence of the enzymatic activity of α -syn has come from studies with purified recombinant protein from bacterial sources and some studies of α -syn expressed in mammalian cells.¹⁹ Overexpression of α -syn in cells elevated the level of Fe(II) present. While this evidence allowed us to

Received: March 21, 2017

Revised: April 28, 2017

Published: April 28, 2017

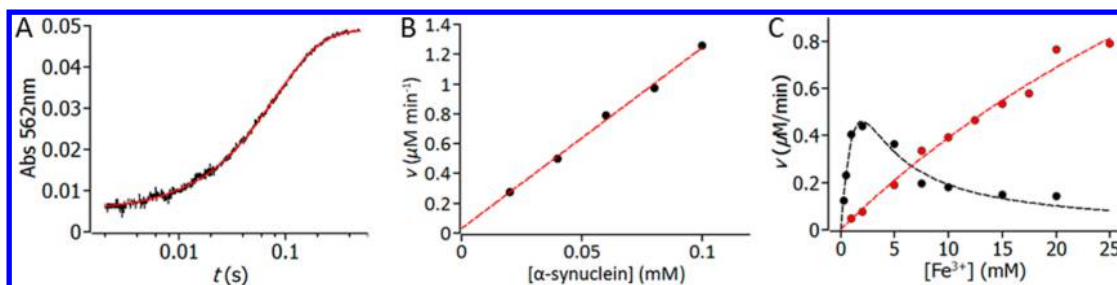


Figure 1. (A) Characterization of the assay used to detect ferrireductase activity. Stopped-flow kinetics monitoring the increase in absorbance at 562 nm upon rapid mixing of 10 μM Fe(II) with ferrozine. The red line is the fit to a single-exponential function. (B) Steady-state kinetics monitoring the formation of Fe(II) in the presence of increasing α -syn concentrations. The red dashed line is the fit to a linear function. (C) Steady-state kinetics of α -syn monitoring the formation of Fe(II) as a function of increasing NADH (red) and Fe(III) (black) concentrations. Solid lines are fits to eq 2 (red) and eq 3 (black).

demonstrate FR activity, it did not provide evidence of the active species of the protein or which domains are necessary for its activity. Such information is of great interest given the recent identification of a tetrameric form of α -syn that has been suggested to be of great importance to understanding the protein's normal role in cells. In this article, we provide more complete evidence of the enzymatic role of α -syn in iron reduction and provide evidence that the active form is a membrane-associated species that is either a dimer or a tetramer.

MATERIALS AND METHODS

Reagents were purchased from Sigma-Aldrich (Poole, U.K.) unless otherwise stated.

Purification of Synuclein Proteins. Using pET expression vectors in BL21 *Escherichia coli* cells, untagged α -syn protein expression was induced at an OD₆₀₀ of 0.5–1.0 with 1 mM isopropyl β -D-1-thiogalactopyranoside for 4 h. Cells were harvested by centrifugation (8000g) and lysed mechanically in 20 mM Tris-HCl with 1 mM EDTA (pH 8.0) (buffer A), 1 mM PMSF, and 50 $\mu\text{g}/\text{mL}$ DNase. Streptomycin sulfate was added to a final concentration of 1% to the lysate solution and then the solution centrifuged at 8000g. Ammonium sulfate (0.295 g/mL) was added to the supernatant [15% (w/v) solution] and stirred at 4 $^{\circ}\text{C}$ for at least 1 h. After centrifugation at 10000g, the pellet was resuspended in 50 mL of buffer A. The semipurified lysate solution was loaded onto a 10 mL Q Sepharose column (Amersham Biosciences). The column was washed with 100 mL of buffer A, followed by a 2 column volume isocratic elution step to 25% buffer A with 1 M NaCl (buffer B). Synuclein proteins were eluted with a broad gradient elution (10 column volumes) from 25 to 50% buffer B (all synuclein proteins eluting at approximately 350 mM NaCl). Sodium dodecyl sulfate–polyacrylamide gel electrophoresis analysis of Q Sepharose fractions was performed, and fractions enriched with synuclein were pooled. Synuclein proteins were collected as flow-through from a PM30 cellulose membrane (Millipore) and then concentrated with a PM10 PES membrane (Millipore). Purified synuclein proteins were dialyzed extensively at 4 $^{\circ}\text{C}$ in Chelex-treated Milli-Q water. The protein concentration was measured by the absorbance at 275 nm with an extinction coefficient of 5600 $\text{M}^{-1} \text{cm}^{-1}$. Mutants of wild-type α -syn were as previously described.²³ The purity of the recombinant protein was confirmed through samples sent to the Mass Spectroscopy Facility at the University of Bristol (Bristol, U.K.). Examples of protein purification are included in Supplementary Figure 1.

Cell Culture. SH-SY5Y (human neuroblastoma) cells were cultured in 45% DMEM with 45% Ham's F12 supplemented with 10% FBS and 1% pen/strep. Cells were maintained at 37 $^{\circ}\text{C}$ and 5% CO₂ in a humidified incubator. The neuronal status of SH-SY5Y cells was monitored by reverse transcription polymerase chain reaction with primers for tyrosine hydroxylase (TH), dopamine transporter (DAT), and vesicle monoamine transporter 2 (VMAT2). Cells were used within 20 passages of the original vial.

Ferrireductase Assay. The ferrireductase activity was measured essentially as previously described,¹⁹ monitoring the concentration dependence of β -nicotinamide adenine dinucleotide hydride (NADH) and ferric ammonium citrate [Fe(III)]. For cell-based assays, cells were grown to 90% confluence, and the cells were scraped from the flask into distilled water and centrifuged immediately at 13K rpm. The resulting membrane pellets were then washed three times in PBS with subsequent centrifugation. The final pellet was then resuspended in PBS with 1% Igepal and incubated on ice for 20 min. The extract was then cleared of debris by centrifugation and the supernatant used for experiments. The protein concentration was determined using a standard Bradford assay (Bio-Rad) and applied to the ferrireductase assay. The cell-based assay used 20 mM MOPS (pH 7), 500 μM NADH, 500 μM ferrozine, and 250 $\mu\text{g}/\text{mL}$ test protein.

Stopped-Flow Fluorescence, Circular Dichroism, and Dynamic Light Scattering Measurements. Absorbance changes were monitored on a thermostated Hi-Tech Scientific stopped-flow apparatus (TgK Scientific, Bradford on Avon, U.K.). Typically, three to five transients were recorded for each reported measurement. Transients were fit as described in the text. Circular dichroism measurements were taken on a thermostated Applied Photophysics Chirascan spectrophotometer (Applied Photophysics, Leatherhead, U.K.) using a high-quality quartz cell with a 0.1 mM path length. Dynamic light scattering (DLS) profiles were recorded as percent volume using a Malvern Zetasizer DLS incubated at 25 $^{\circ}\text{C}$.

Nanodisc Preparation. L- α -Phosphatidylcholine from soybean (Sigma, catalog no. P3644-25G) was resuspended in 20 mM Hepes (pH 7.5) and 100 mM NaCl to give a 10 mM suspension of liposomes. This suspension was extruded for 15 passes through a 100 nm membrane (Sigma, catalog no. WHA800309) using a mini extruder (Avanti Polar Lipids). Hydrolyzed SMA polymer (3:1 styrene:maleic acid ratio) was a gift from K. Edler (Department of Chemistry, University of Bath). The polymer was solubilized in 20 mM Hepes (pH 7.5) and 100 mM NaCl at 6% (w/v). For the preparation of

nanodiscs, a 1:3 ratio of 6% (w/v) SMA to 10 mM extruded liposomes was incubated at 18 °C for 2 h. Nanodisc preparations [1.5% (w/v) SMA and 7.5 mM lipid] were analyzed by dynamic light scattering (DLS) for quality control. We note that the size of the nanodiscs as assessed by DLS corresponds with previous reports based on NMR studies (see below).

RESULTS

α -syn Is a Bona Fide Ferrireductase. We used an established assay to detect the formation of Fe(II), based on the optical absorption of 3-(2-pyridyl)-5,6-bis(phenylsulfonic acid)-1,2,4-triazine (ferrozine) at 562 nm, which increases upon binding of Fe(II). The assay is highly specific for the presence of Fe(II) with essentially no change in the presence of Fe(III). This indirect assay has the potential to be a confounding factor in the analysis of ferrireductase activity. That is, if the rate of binding of Fe(II) to ferrozine is slow, or at least slower than the activity of the enzyme, then Fe(II) binding becomes the rate-limiting step. We have therefore monitored the rate of binding of Fe(II) to ferrozine, assessed by the absorption change at 562 nm in a stopped-flow apparatus. An example stopped-flow transient is shown in Figure 1A. The data fit to a single-exponential function:

$$\Delta A_{\text{obs}} = A \exp(-k_{\text{obs}}t) \quad (1)$$

where ΔA_{obs} is the change in the observed absorption of ferrozine with respect to time (t) and A is the amplitude of the kinetic transient giving an observed rate constant (k_{obs}).

The fact that the data fit to a single exponential suggests only a single physical step in the binding interaction. The observed rate extracted from eq 1 (k_{obs}) is $12.29 \pm 0.1 \text{ s}^{-1}$. The observed rate of binding of Fe(III) to ferrozine is therefore at least 3 orders of magnitude faster than the values of k_{cat} we observe for α -syn (*vide infra*), so at least for α -syn, the assay reflects enzyme turnover and Fe(II) formation. We cannot comment on the mechanistic step or steps we access with this assay, however, other than it most likely reflects the slowest step in the turnover of Fe(III) to Fe(II).

The key experimental test for enzyme activity is the observation of a linear relationship between the observed rate of turnover and the enzyme concentration. Figure 1B shows the concentration dependence of α -syn on the observed rate of Fe(II) formation. We found that the data fit to a simple linear function that intersects both axes essentially at zero. That is, the rate of Fe(II) formation increases linearly with the concentration of α -syn, suggesting that the observed ferrireductase reaction is truly enzyme-catalyzed by α -syn. As an additional step of verification, we compared α -syn ferrireductase activity with and without boiling of the protein. Boiling of the protein for 1 h completely inactivates the protein (Supplementary Figure 2).

Figure 1C shows the concentration dependence of Fe(III) and NADH on the rate of Fe(II) formation in the presence of 100 μM α -syn. For NADH, we found that the data can be adequately fitted to the Michaelis–Menten equation, giving a k_{cat} of $(2.93 \pm 0.95) \times 10^{-2} \text{ min}^{-1}$ and a K_{m} of $65.2 \pm 26.9 \text{ mM}$. For Fe(III), the data do not fit to a simple Michaelis–Menten relationship (eq 2) but instead show an initial increase in observed rate and then a decrease at higher Fe(III) concentrations. The data fit to a model that includes substrate inhibition (eq 3)

$$v = \frac{k_{\text{cat}}[E]_0[S]}{K_{\text{m}} + [S]} \quad (2)$$

$$v = \frac{k_{\text{cat}}[E]_0[S]}{K_{\text{m}} + [S] \left(\frac{1 + [S]}{K_i} \right)} \quad (3)$$

where K_i is an inhibition constant. Fitting the data to eq 3 gives a k_{cat} of $(3.66 \pm 4.9) \times 10^{-2} \text{ min}^{-1}$, a K_{m} of $7.0 \pm 10.5 \text{ mM}$, and a K_i of 0.57 ± 0.85 . The magnitude of k_{cat} is relatively small. This could point to either a high barrier to transition-state formation, a very slow rate-limiting step in the overall enzymatic turnover (including substrate binding, chemical turnover, and product release), or a small fraction of the total protein being in a catalytically competent conformation. The observation of a large K_{m} value in the millimolar region for both NADH and Fe(III) suggests the ternary reactive complex is weakly bound.

For such a two-substrate system, one would like to take the steady-state measurements at a saturating concentration of one substrate. This is not possible in this case because the $K_{\text{M}}^{\text{app}}$ values are large, and saturating these would place the concentrations beyond the solubility limits of the substrates. We therefore prefer to interpret the relative trends in our data rather than the absolute magnitudes.

α -syn Variants Suggest a Pathological Link to Ferrireductase Activity. We explored the steady-state kinetics of a range of key, pathologically relevant α -syn variants shown in panels A and B of Figure 2 with the resulting extracted values from the fits given in Table 1. Figure 2C shows a summary of our findings with respect to the structural location of the α -syn variants.

(i) A53T and E46K are mutations in the SPCA gene associated with inherited forms of PD. The K_{m} was relatively unaffected for NADH with a decrease for Fe^{3+} , but k_{cat} was significantly reduced in both cases. (ii) H50Q is a mutation identified from inherited cases of PD. The H50 site could influence metal binding and is associated with increased levels

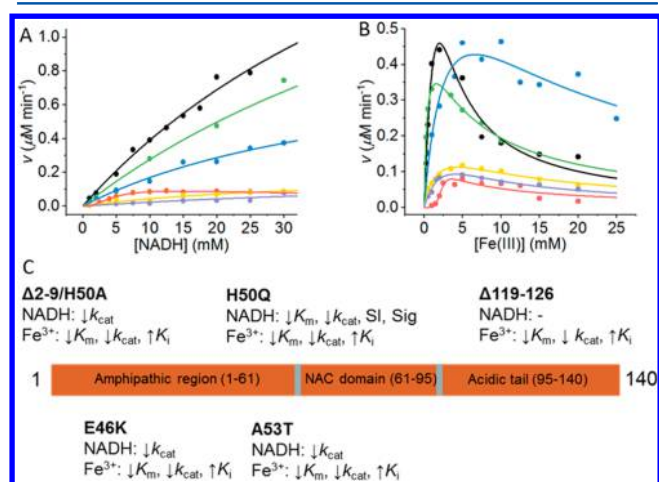


Figure 2. Steady-state kinetics of α -syn showing concentration dependencies with (A) NADH and (B) Fe^{3+} . Solid lines are fits to models described in Table 1, with the corresponding extracted parameters: black for wild-type synuclein, coral for H50Q, blue for A53T, yellow for E46K, green for $\Delta 119-126$, and lilac for $\Delta 2-9/\text{H50A}$. (C) Summary of the effect of different variants on the steady-state kinetics of α -syn.

Table 1. Extracted Parameters from Steady-State Data Fitting Shown in Figure 2

	V_{\max}^{app} ($\mu\text{M}/\text{min}$)	K_M^{app} (mM)	$k_{\text{cat}}^{\text{app}}$ (min^{-1})	$k_{\text{cat}}^{\text{app}}/K_M^{\text{app}}$ ($\text{min}^{-1} \text{mM}^{-1}$)	K_i^{app} (mM)	n
NADH Concentration Dependence						
$\alpha\text{-syn}^a$	2.93 ± 0.95	65.2 ± 26.9	2.93×10^{-2}	4.50×10^{-4}	—	—
$\Delta 119\text{--}126$	2.5 ± 0.7	80.9 ± 32.8	2.5×10^{-2}	3.1×10^{-4}	—	—
H50Q ^b	0.12 ± 0.02	5.0 ± 0.8	0.12×10^{-2}	2.4×10^{-4}	63.3 ± 40.0	2.1 ± 0.3
A53T	0.94 ± 0.24	45.6 ± 17.5	0.94×10^{-2}	2.0×10^{-4}	—	—
E46K	0.19 ± 0.03	32.9 ± 10.4	0.19×10^{-2}	5.7×10^{-5}	—	—
$\Delta 2\text{--}9/\text{H50A}$	0.20 ± 0.15	73.6 ± 82.5	0.20×10^{-2}	2.7×10^{-5}	—	—
Fe^{3+} Concentration Dependence						
$\alpha\text{-syn}^c$	3.66 ± 4.9	7.0 ± 10.5	3.66×10^{-2}	5.2×10^{-3}	0.57 ± 0.85	—
$\Delta 119\text{--}126$	0.55 ± 0.02	0.46 ± 0.1	0.55×10^{-2}	1.12×10^{-2}	5.43 ± 0.53	—
H50Q ^b	0.13 ± 0.06	2.3 ± 0.4	0.13×10^{-2}	6×10^{-6}	7.1 ± 5.8	5.4 ± 3.0
A53T	0.84 ± 0.24	3.14 ± 1.6	0.84×10^{-2}	2.7×10^{-5}	13.8 ± 7.5	—
E46K	0.21 ± 0.03	1.9 ± 0.57	0.21×10^{-2}	1.1×10^{-6}	10.0 ± 3.1	—
$\Delta 2\text{--}9/\text{H50A}$	0.25 ± 0.1	3.39 ± 1.8	0.25×10^{-2}	7.2×10^{-6}	5.0 ± 2.8	—

^aMM kinetics. ^bSubstrate inhibition. ^cCooperative binding and substrate inhibition. In all cases, the simplest model was selected on the basis of the extraction of realistic fitting parameters and statistical quality of the fit.

of oligomerization and fibril formation of $\alpha\text{-syn}$. We find a significantly reduced K_m but also a significantly lower k_{cat} for both NADH and Fe(III) . We observed a complex kinetic relationship for both substrates, exhibiting both a sigmoidal relationship, suggestive of cooperativity, and substrate inhibition. We fitted these data to the function

$$v = \frac{k_{\text{cat}}[E]_0[S]^n}{K_m^n + [S]^n + \left(\frac{1 + [S]^n}{K_i}\right)} \quad (4)$$

Where n is a constant reflecting cooperativity.

Such a relationship is paradigmatic of the activity of other enzymes, for example, aspartate transcarbamylase.²⁴ These data therefore suggest either induction or unmasking of a cooperative relationship. Potentially, these data may indicate that there are multiple discrete binding geometries for both NADH and Fe(III) in the H50Q variant and certainly that this metal binding site is key to the formation of the optimal reactive geometry with both substrates. (iii) $\Delta 2\text{--}9/\text{H50A}$ is a deletion of the N-terminus of the protein that has been shown to be the site of high-affinity copper²⁵ binding and also removes the H50 residue from any potential role in metal binding. The K_m is relatively unaffected for NADH with a decrease for Fe(III) , but k_{cat} is significantly reduced in both cases. These data therefore suggest that the N-terminus is not significantly involved in the binding of either NADH or Fe^{3+} but is required for full enzymatic activity. Potentially, copper binding increases the propensity to form a catalytically competent conformation. (iv) $\Delta 119\text{--}126$ represents the removal of a potential low-affinity divalent metal binding site. The k_{cat} and K_m for NADH are hardly affected by this variant, but the k_{cat} and K_m are significantly affected for Fe(III) . The mutant data suggest this site is not involved in NADH binding or oxidation but is key to reduction and binding of the Fe(III) .

Differences in k_{cat} between the mutants could be explained either by different conformational states of the proteins or by a smaller or larger proportion of a catalytically active conformation, or alternatively as a consequence of the specifics of the two-substrate mechanism. However, differences in equilibrium constants K_m and K_i are not dependent on the proportion of the catalytically active species but instead on differences in binding geometry, particularly for the reactive complex. The fact that we observe significant changes in k_{cat} ,

K_m , K_i , and the induction of cooperative-like behavior (H50Q) argues strongly for structurally different forms of the mutants studied. Crucially, this does not rule out a fraction of the recombinant protein being active, and we explore this possibility below. We note that we do not suggest that the observed kinetic differences are the sole driver of the disease state, because clearly aggregation propensity is a key factor. However, the key finding is that these data demonstrate that these disease variants can significantly alter the kinetics of FR activity.

CD Analysis. We have explored whether there is evidence of $\alpha\text{-syn}$ mutant conformational change through comparative far-UV circular dichroism (CD) spectroscopy shown in Figure 3.

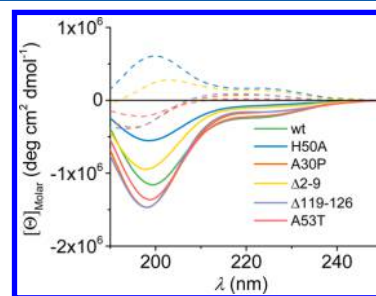


Figure 3. Far-UV CD spectra of recombinant $\alpha\text{-syn}$ and variants. Solid lines are spectra corresponding to each variant, and the dashed lines are the corresponding difference spectra for each variant with respect to wild-type $\alpha\text{-syn}$.

The far-UV CD spectra of each mutant, as with wild-type (WT) $\alpha\text{-syn}$, are indicative of a protein with an extremely high degree of unstructured content. However, the spectra show clear differences; the dashed line in Figure 3 shows the difference spectra of each mutant compared to the WT $\alpha\text{-syn}$. In the case of A30P, A53T, and $\Delta 119\text{--}126$ $\alpha\text{-syn}$, the difference spectra are extremely characteristic of unstructured content with a negative peak at ~ 198 nm and a small broad and positive peak at ~ 220 nm. The difference spectra for H50A and $\Delta 2\text{--}9$ are complex and do not present an obvious structural interpretation. Far-UV CD data are limited in their ability to quantitatively interpret unstructured content. However, our data suggest there is at least some conformational difference arising from the mutations studied. These data therefore

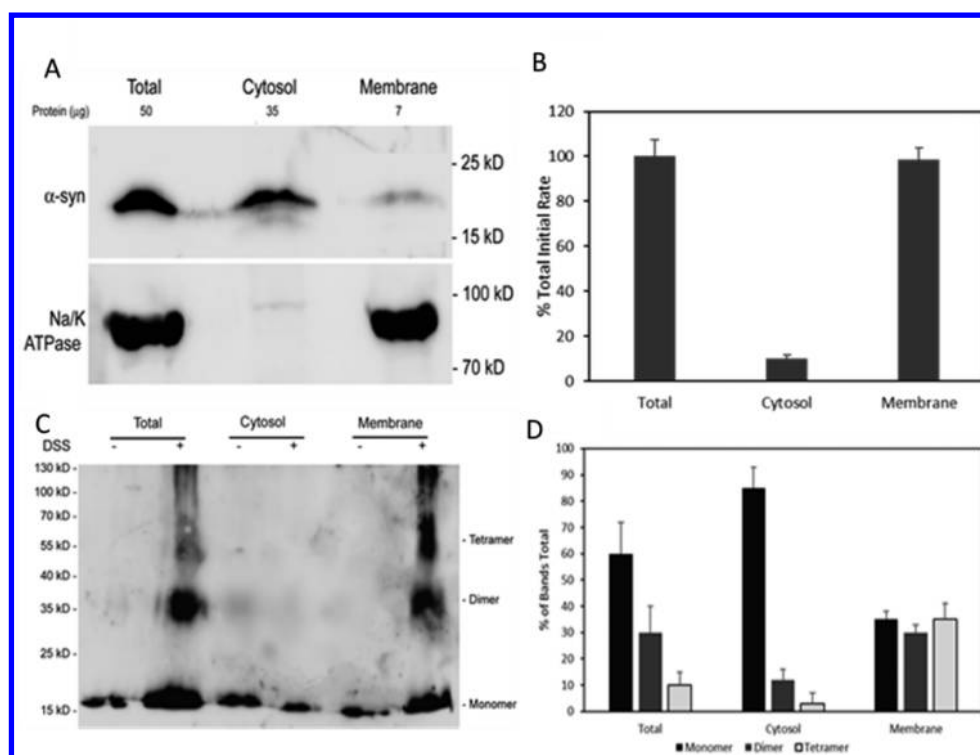


Figure 4. Identifying the structural form of FR active α -syn. (A) Western blot of equivalent volumes of protein extracts prepared from SH-SY5Y cells overexpressing α -syn, separated into cytosolic and membrane fractions. The levels of α -syn and Na/K ATPase were assessed by immunodetection. (B) The fractions from panel A were then assayed for FR activity, quantified on the basis of differences in the observed initial rate of ferrozine absorption change in the presence of 500 μ M Fe(III). (C) Western blot detecting cross-linked α -syn from the SH-SY5Y fractions from panel A. (D) Relative amount of different α -syn oligomers from panel C were detected by densitometry. Data are the means and standard errors of the mean for four experiments.

provide a rationale for the different equilibrium constants, suggesting that the mutant structures affect the formation of the reactive ternary complex and potentially part of the differences in the extracted k_{cat} values. The absolute magnitude of the k_{cat} value for the mutants studied might vary on the basis of an increasing or decreasing concentration of a specific structural form. In particular, given the unordered structure of WT and mutant proteins, there is significant scope for a broad equilibrium of conformational states, and we explore this possibility below.

Identification of the Active Structural Form. We have previously shown the FR activity of α -syn in cells.¹⁹ The cell-based system has a distinct advantage over pure recombinant protein in that it is possible to assess the localization of an active form of the protein. Ferrireductases are frequently membrane-associated.^{26,27} α -syn is also known to have a membrane association.²⁸ Therefore, we investigated the possibility that α -syn present in the membrane is more FR active than that present in the cytosol. A crude membrane extract was made from SH-SY5Y cells overexpressing α -syn. The level of α -syn present was compared between membrane and cytosolic fractions using Western blotting (Figure 4a). The majority of detectable protein was present in the cytosol. However, a small percentage was membrane-associated. The quality of membrane isolation was verified using Western blotting of the Na/K ATPase antiporter. The level of FR activity was then assessed for the membrane and cytosolic fractions and compared to activity in an unfractionated cell extract (total). Virtually all of the FR activity was present in the membrane rich fraction despite containing only a small fraction

of the total cellular α -syn (Figure 4b). The implication is that the majority of FR activity present in α -syn-overexpressing cells is located in the membrane fraction of SH-SY5Y cells. As a further step to verify that this activity was from α -syn, we used a specific monoclonal antibody to immunoprecipitate (IP) α -syn from the membrane fractions. Ferrireductase activity was associated with only anti- α -syn IP material but not IP material when a control antibody was used (Supplementary Figure 3).

The localization of a small fraction of α -syn to the cell membrane and the association of this fraction with the majority of FR activity infer the structural identity of the most active species of α -syn. There has been a suggestion that α -syn can be expressed in cells as a helical tetramer.¹¹ Identification of tetrameric α -syn can be achieved by cross-linking of the protein within cell extracts. Total membrane and cytosolic fractions of SH-SY5Y cells overexpressing α -syn were cross-linked, and α -syn was identified using Western blotting (Figure 4c). After fractionation, virtually all dimeric and tetrameric species of α -syn were present in the membrane fraction with almost solely monomeric protein in the cytosolic fraction. Also, in comparison to the total fraction, the level of tetrameric protein is clearly enriched. This strongly supports the notion that the active species is the tetrameric form of α -syn. Potentially, the relatively small activity of the cytosolic fraction can be attributed to the presence of a minor amount of enzymatically active oligomeric α -syn present in the cytosol (Figure 4d).

Our finding that the major species showing ferrireductase activity is associated with the membrane affords the opportunity to explore FR activity in the structural form found in the membrane, which we infer is a helical tetramer as

described previously.^{11,29} To that end, we have prepared protein extracts of the membrane fraction from cell lines overexpressing α -syn and a range of mutants. We assessed the kinetics of $\Delta 2-9$, $\Delta 1-100$, H50A, $\Delta 2-9$ /H50A, and $\Delta 2-9$ /H50A/ $\Delta 111-140$ to explore whether selected mutants have an additive effect. The observed activity with respect to Fe(III) concentration is shown in Figure 5 at essentially the same

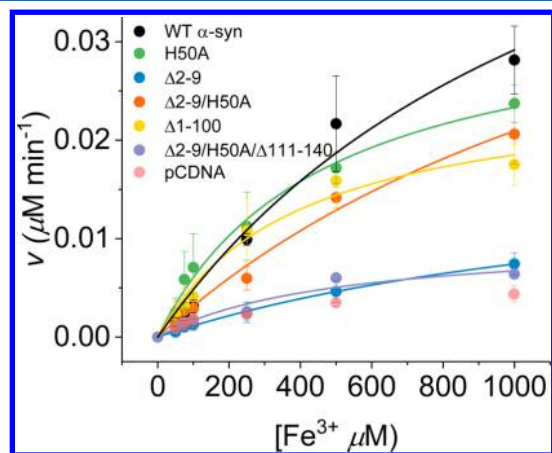


Figure 5. Steady-state kinetics of protein extracts from α -syn-expressing cell lines showing concentration dependencies for Fe(III). The membrane fractions from the cells were prepared. Protein concentrations were 250 $\mu\text{g}/\text{mL}$. The extracted kinetic parameters are listed in Table 2.

concentration of extracted total protein (250 $\mu\text{g}/\text{mL}$) and the extracted kinetic parameters listed in Table 2. The concen-

Table 2. Extracted Parameters from Steady-State Data Fitting Shown in Figure 5

	V_{\max} ($\mu\text{M}/\text{min}$)	K_m (mM)
α -syn	0.066 ± 0.02	1.28 ± 0.59
H50A	0.034 ± 0.003	0.46 ± 0.08
$\Delta 2-9$	0.018 ± 0.002	1.53 ± 0.19
$\Delta 2-9$, H50A	0.056 ± 0.014	1.68 ± 0.6
$\Delta 1-100$	0.026 ± 0.003	0.40 ± 0.1
$\Delta 2-9/\Delta 111-140/\text{H50A}$	0.01 ± 0.002	0.5 ± 0.2
pCDNA (negative control)	0.005 ± 0.001	255.9 ± 51

tration range of Fe(III) we explore is necessarily more restricted than that in our recombinant protein studies because of the inherent capacity of high iron concentrations to cause cell components to precipitate. We therefore focus on the general trends that are apparent from the fitting shown in Figure 5. From the extracted kinetic parameters, it is apparent that the variant enzymes show significant differences in the overall rate of turnover (assessed by V_{\max} , not k_{cat} , because we do not know the precise enzyme concentration). Perhaps most importantly and consistent with our recombinant protein studies mentioned above, there is a clear difference in K_m , suggesting structurally different forms of the reactive complex geometry between the mutants. Notably, the triple mutant, $\Delta 2-9/\text{H50A}/\Delta 111-140$, has the lowest activity, which approaches that of the empty vector control cell extract (Table 2). While we have fitted the control vector to the Michaelis–Menten equation, this is more for comparison because we do not expect these data to obey saturation behavior with respect to Fe(III) concentration and the fitting

simply serves to illustrate the extremely weak signal arising from potentially other membrane-associated ferrireductases.

We wished to explore whether an artificial membrane environment could be used to drive the formation of helical α -syn and whether this is sufficient to induce FR activity. Recent studies have demonstrated that nanodiscs, defined artificial membrane environments, can be used to induce α -syn to form helical multimers.³⁰ As these nanodiscs had a protein scaffold, they are not compatible with many protein biophysical techniques because of background “noise” from the discs themselves. For this reason, we used styrene–malate polymer scaffolds to make nanodiscs. These have been shown to be compatible with a range of optical biophysical measurements, including CD and dynamic light scattering (DLS). We have incubated 5 μM recombinant α -syn in the presence of 50 μM lipid nanodiscs diluted from 7.5 mM nanodiscs prepared as described in Materials and Methods.³¹ To assess α -syn incorporation and folding into the nanodiscs, we have monitored the far-UV CD spectra and the dynamic light scattering (DLS) profile as shown in panels A and B of Figure 6, respectively. We find that when α -syn is incubated with the

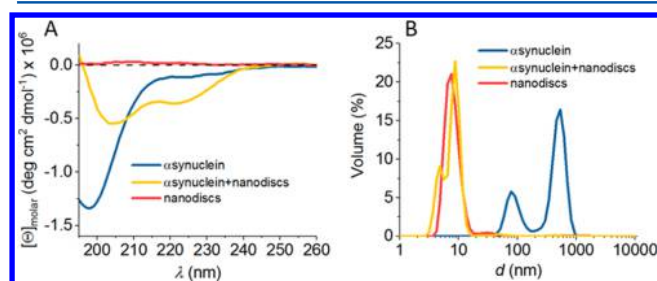


Figure 6. Incorporation of α -syn into nanodiscs. (A) Far-UV CD spectra of recombinant free α -syn and bound to nanodiscs. (B) DLS (volume) profiles of recombinant free α -syn and α -syn bound to nanodiscs.

nanodiscs, the far-UV CD data become consistent with a protein that is predominantly helical, showing negative ellipticity centered at ~ 205 and 222 nm. From Figure 6B, the DLS data show that the recombinant α -syn is initially at a high hydrodynamic diameter, indicative of a low- n aggregated species. From Figure 6B, the nanodiscs give a diameter of ~ 10 nm, consistent with previous reports of nanodiscs of this composition.³² Incubation with the nanodiscs disaggregates this species, suggesting that the helical species incorporated into the nanodiscs is not an aggregate or large multimer. On assaying the nanodisc– α -syn species with the FR assay described above, we found no detectable FR activity.

DISCUSSION

The function of many proteins associated with neurodegenerative diseases remains unresolved and frequently controversial. This is especially true for α -syn, a protein heavily implicated in a range of diseases collectively termed the synucleinopathies. Our interest in the identification of a function for these proteins underlies the possibility that a loss or alteration of this function may play a causal role in the pathology. α -syn has been linked to altered metal metabolism. In PD, there is especially a well-known change in iron metabolism.^{21,22,33} Therefore, a function for α -syn related to iron is of great significance.

We previously provided evidence that α -syn acts as an FR.¹⁹ An FR plays a vital role in a cell that requires Fe(II) as it allows the regeneration of this form of iron that is vital for many cellular activities. In dopaminergic neurons, Fe(II) is required as a cofactor for tyrosine hydroxylase, which is the key, rate-limiting enzyme in the synthesis of dopamine. We showed that α -syn can increase cellular Fe(II) levels and that cells with increased α -syn levels have increased FR activity. Additionally, we showed using recombinant protein that α -syn exhibits FR activity when purified. In this report, we provide a much expanded concentration range of substrates and explore the concentration dependence of both NADH and Fe(III). These data allow greater insight into the FR activity of α -syn, providing evidence of its bona fide enzymatic status, the role of substrate inhibition, and its dependence on NADH for activity. Most significantly we have provided evidence from cell studies that the active form of α -syn is a tetramer. This supports recent studies that have pointed out the structured (helical) tetrameric species as the more important subspecies *in vivo*.^{11,34,35}

Potentially, the α -syn ferrireductase activity is relatively low because only a small fraction of the protein exists in a conformation that is catalytically competent. This is highly likely as demonstrated by the cell extract data where activity is associated with a subspecies isolated from a membrane. It is important to note that when considering the recombinant protein the amount of tetramer present is much smaller than the amount that can be detected in cells [$\sim 2\%$ of the total (data not shown)]. Alternatively, the activity could be relatively low because α -syn is primarily disordered in solution and therefore lacks a well-defined reactive geometry in the active site. This is less likely if the active species is a tetramer, as this has been shown to have a high helical content.^{11,35} However, we note that the magnitude of K_m is independent of enzyme concentration, so even if a small fraction of the protein is catalytically active, the enzyme itself still displays a high K_m in relation to the cellular concentrations of both NADH and Fe³⁺. Clearly, however, local concentrations of substrates can be very much higher at the microscopic level.

From our recombinant protein studies, we have found that the data fit to a model that includes substrate inhibition. The observation of substrate or product inhibition is relatively common and usually implies that the active site geometry can accommodate the substrate in more than one unique conformation, and at least some of these conformations are not catalytically competent. Typically, one observes the K_i value to be rather larger than the K_m because the affinity of binding of the substrate in the catalytically competent configuration is anticipated to be higher than that in noncompetent configurations. However, we find that for Fe(III) there is a “turned-out” relationship for K_m and K_i , with the K_i being rather smaller than the K_m . A consequence of this unusual relationship is that wild-type α -syn exhibits a very sharp concentration dependence in activity as shown in Figure 2. We suggest that the relationship between K_m and K_i could be functionally important in governing the dose–response relationship of the intracellular pool of Fe(III) and Fe(II). That is, the ferrireductase activity of α -syn will become significant only over a very narrow concentration range of available Fe(III).

Our data suggest that the enzymatically active form of α -syn is a membrane-associated species, most likely to be the helical tetrameric species described recently.¹¹ However, we find that α -syn helicity is not itself sufficient to induce FR activity, with our artificial nanodisc– α -syn preparations showing no FR

activity. Instead, the FR activity is primarily associated with a structural form found in non-artificial cell membranes. These data argue that it is the precise quaternary structure of α -syn oligomers that defines an active site compatible with FR activity. This finding suggests a possible mechanism by which α -syn acts as a ferrireductase. FR activity very commonly proceeds through a flavin intermediate, with hydride transfer from NAD(P)H, reducing the flavin and with subsequent reduction of Fe(III).³⁶ Four-helix bundle architectures have been shown to bind flavin moieties and act as flavoenzymes.³⁷ A realistic possibility for the FR activity of α -syn is therefore that the helical tetrameric assembly is capable of (likely weakly) binding flavin and using this species as a coenzyme to catalyze the reduction of Fe(III). This notion is satisfying because it would explain why we find the large majority of the FR activity associated with the membrane fraction where one expects the native helical tetramer to be prevalent. Moreover, this hypothesis would be consistent with the need for the helical oligomeric assembly to adopt a precise quaternary arrangement.

Kinetic assays with purified membrane protein fractions show saturation kinetics, and the mutant proteins we have monitored show significant differences in both the overall turnover rate and the K_m . These data therefore are consistent with our recombinant protein assays in demonstrating that α -syn variants can affect FR activity. The observed differences in overall turnover and K_m might point to a difference in the active site architecture. With that said, we have shown previously that the $\Delta 2-9/H50A$ mutant and deletions of the C-terminus alter membrane association.³⁸ The difference in the observed activity might therefore have an origin in an altered concentration of a catalytically competent structural form, i.e., membrane-associated versus soluble. Another possibility is that the mutations alter the rate of tetramer formation. A recent paper has identified the KTKEGF repeat regions through the protein sequence as being essential for tetramer formation. In contrast, deletions of 10-amino acid residue stretches (e.g., $\Delta 2-10$) had no effect on tetramer formation.³⁴ The implication is that changes in activity seen in our mutants are unlikely to be due to altered tetramer formation but instead are likely due to changes in membrane association. However, the key finding is these data support our notion that the FR active species is membrane-associated and by inference the recently identified helical tetramer structural form.

The FR activity of α -syn is potentially important in understanding its pathological role, particularly because iron homeostasis is known to be important in PD. Herein, we sought to explore steady-state kinetics of α -syn FR activity with a range of disease-associated variants and by examining the nature of the catalytically active species. By exploring extended concentration ranges for the two substrates [NADH and Fe(III)] we found that turnover with NADH follows classical Michaelis–Menten kinetics, but with Fe(III), there is very significant substrate inhibition. Most importantly, the kinetics of enzyme turnover, notably the inhibition and Michaelis constants, were significantly different with the disease variants. These data suggest that known disease variants of α -syn can “tune” the FR activity. The observed rates of enzyme turnover are slow, and we suggest this is because only a fraction of the enzyme is in a catalytically competent conformation, lacks a necessary cofactor, or both. Whole cell assays point to the FR active α -syn being a membrane-associated tetrameric species, which we suggest is the recently identified helical tetrameric form. By using artificial membrane constructs (nanodiscs), we

found that inducing α -syn to form a helical conformation is not itself sufficient to induce FR activity, suggesting that the absolute precision of the quaternary arrangement of helical α -syn monomers is necessary to define the catalytically active species. We go on to suggest that a rational mechanism for the FR activity of α -syn might be the binding of a flavin coenzyme by the helical tetrameric α -syn, which then mediates the transfer of an electron from NADH to Fe(III) as in other ferrireductases.

■ ASSOCIATED CONTENT

● Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.biochem.7b00257.

Supplementary Figures 1–3 (PDF)

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Funding

The authors thank the UK charities BRACE and Alzheimer's Research UK (ARUK-PG2012-1) for funding the project.

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

The authors thank Mr. Morgan Roberts for technical assistance during the preparation of the nanodiscs.

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